

Au-Catalyzed Pentannulation Reaction of Propargylic Esters Occurring at C(sp³)–H Site

Roohollah Kazem Shiroodi, Masumi Sugawara, Maxim Ratushnyy, Douglas C. Yarbrough, Donald J. Wink, and Vladimir Gevorgyan*

Department of Chemistry, University of Illinois at Chicago, 845 West Taylor Street, Chicago, Illinois 60607-7061, United States

Supporting Information

ABSTRACT: A gold-catalyzed cascade cyclization reaction of easily accessible propargylic esters to cyclopentenones has been developed. This transformation features an unprecedented pentannulation reaction of propargylic esters which occurs at an unactivated $C(sp^3)$ -H site to efficiently produce functionalized mono-, bis-, and tricyclic cyclopentenones.



Readily available propargylic esters are versatile substrates for transition-metal-catalyzed transformations into a wide array of valuable acyclic, cyclic, and heterocyclic motifs.¹ Mechanistically diverse π -acidic metal-catalyzed pentannulation reactions of these substrates, possessing carbonyl or epoxide functionalities at the alkyne moiety, highlight their rich reactivities (Scheme 1).² Thus, Sarpong and co-workers reported that Pt-catalyzed Rautenstrauch rearrangement of propargylic ester 1 generates the metal carbene a, which upon insertion into the C(sp²)–H bond produces cyclopentene 2.^{2a} On the other hand, the Toste group showed that Au-catalyzed 1,2-acyloxy migration of substrate 3 produces nonclassical cationic intermediate b, in which the generated cation is stabilized by the indole moiety. A ligand-controlled enantioselective imino-Nazarov cyclization of





Scheme 2. Initial Reaction Design for the Synthesis of Fully Substituted Heterocycles







the latter produces the cyclopentenone-fused indoline 4.^{2b} In addition, Pale and Blanc et al. have recently reported that alkynyl epoxide 5 in the presence of a Au catalyst generates dienone c via a 1,2-acyloxy migration/oxygen delivery from an epoxide moiety, which upon Nazarov cyclization delivers the cyclopentenone 6. All of the above transformations involve initial 1,2-acyloxy migration and furnish the pentannulated products either by cyclization at the $C(sp^2)$ -H site (a) or via Nazarov annulation (b and c). Herein, we wish to report a new mode of pentannulation reaction of propargylic ester 7 in the presence of a gold catalyst,³ where a cascade sequence of reactions involving a 1,3-acyloxy migration generates intermediate d, which upon cyclization produces fused pivalated cyclopentenone 8. This transformation features an unprecedented involvement of an unactivated $C(sp^3)$ -H site⁴ in the pentannulation reaction of propargylic esters.

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Table 1. Substrate Scope in Pentannulation Reaction of Propargylic Esters



We have previously shown that the tertiary propargylic phosphates 9 (R = H) possessing a ketone moiety, in the presence of a copper catalyst, undergoes a facile migratory cycloizomerization reaction via the allene **e** to produce functionalized furan 10 (Scheme 2).⁵ We hypothesized that by using the quaternary propargylic system 9 (R= alkyl, Ar), the

transition-metal-catalyzed double migratory process⁶ with the intermediacy of the tetrasubstituted allene **e** would be feasible⁷ to produce the fully substituted furan **10**. To this end, the reactions of propargylic ester **7a**, possessing a five-membered ring toward the fused furan **11a** (via a 1,3-pivaloxy migration/ring expansion) was examined (Scheme 3).⁸ Also in the presence of a silver

Scheme 4. Mechanistic Studies



Scheme 5. Proposed Mechanism



Scheme 6. Deuterium-Labeling Experiment



catalyst this transformation occurs to produce expected fused furan 11a; the conditions for more efficient and general transformation have not been identified yet.⁸ It was found that in the presence of a cationic gold catalyst the 1,3-pivaloxy migration^{6b} occurred to give the expected allene 12a in 56% yield, which, however, did not further cyclize into 11a even under elevated temperatures.⁸ Surprisingly, employing the $(C_6F_5)_3AuCl/AgOTf$ catalyst system⁹ produced the fused cyclopentenone 8a¹⁰ in 73% isolated yield!

Inspired by the unexpected reactivity of propargylic esters and by the importance of the obtained products,¹¹ we explored the scope of this cascade pentannulation transformation (Table 1). Thus, both substrates 7a and 7b possessing pivaloxy and acetoxy migrating groups reacted smoothly under the reaction conditions to produce fused cyclopentenones 8a and 8b in good yields (entries 1, 2). Ortho-fluorinated phenyl-substituted cyclopentenone 8c was obtained in 54% yield (entry 3). Pentannulation reactions of substrates 7d and 7e, possessing *meta*-trifluoromethyl phenyl and 2-naphthyl groups, occurred uneventfully affording products 8d and 8e in reasonable yields (entries 4, 5). It was found that propargylic esters 7f-k, containing various groups at the para position of the phenyl moiety, are competent reactants in this cyclization. Thus, substrates 7f and 7g, possessing methyl and chlorine groups, worked well to give rise to 8f and 8g in 50% and 68% yields, respectively (entries 6, 7). Likewise, propargylic esters 7h-k, possessing electron-deficient substituents, allowed for the efficient synthesis of fused cyclopentenones 8h-k (entries 8-11). Notably, propargylic pivalate derivatives of various heterocycles, including furan 7l, thiophene 7m, and benzofuran 7n cyclized in good to excellent yields (entries 12–14). Cyclohexylcontaining substrate 70 underwent this cascade reaction to produce alkylated cyclopentenone 80 in a modest yield (entry 15). The benzannulated substrate 7p cyclized into the tricyclic cyclopentenone 8p. We wondered whether the observed cascade pentannulation reaction is specific to cyclopentane-containing substrates or is more general.¹² To this end, we examined a potential cyclization of the acyclic substrate 7q (analogue of 1), which upon Pt-catalysis is known to cyclize at the $C(sp^2)$ -H site to produce 2 (Scheme 1, eq 1).^{2b} Remarkably, pentannulation reaction of 7q at a lower temperature led to the exclusive cyclization at the $C(sp^3)$ -H bond to produce the monocyclic cyclopentenone 8q.

After establishing the scope of this gold-catalyzed pentannulation reaction, we performed some preliminary mechanistic studies (Scheme 4). First, subjecting the allenyl ketone 12a to the reaction conditions did not lead to the cyclized product 8a even under forcing conditions (eq 1). This observation rules out possible involvement of allene intermedite 12a in this reaction. Next, to determine whether this cascade reaction involves a skeletal rearrangement, which is common for the gold-catalyzed reactions,¹³ we performed the ¹³C-labeling studies. Thus, subjecting the substrate 13, labeled at the C2 of the alkyne moiety, to the reaction conditions afforded cyclopentenone 14 with the labeled carbon atom at the ketone moiety (eq 2). In addition, pentannulation reaction of 15, possessing the labeled carbon at the carbonyl group of alkynyl ketone, produced 16 with the 13 C-labeled carbon residing at the olefinic C4 position (eq 3). Hence, these observations imply that this migratory cyclization reaction does not involve a skeletal rearrangement of the ynone moiety but involves a 1,3-acyloxy migration/1,3-oxygen transfer/ cyclization cascade.

Based on the above observations, we propose the following plausible mechanism for this novel cascade transformation (Scheme 5). First, the π -philic gold catalyst activates the alkyne moiety of 7 to form **A**, thus triggering equilibrium to produce the 1,3-transposed¹⁴ isomer **B**.¹⁵ The following *6-exo-dig* cyclization of the latter would occur by attacking the nucleophilic oxygen of the ester group at the activated alkyne moiety to produce **C**.¹ According to **path a**, a 1,2-H in **C** via the transition state **D** gives enone **E**, which further cyclizes to the fused cyclopentenone **8** upon nucleophilic attack of the vinyl gold species at the cationic center. Alternatively, the proton loss α to the spiro-center of **C** would occur to open the ring and form diene **F** (**path b**). Protiodemetalation of the latter would generate the activated dienone **G**, which upon Nazarov cyclization¹⁶ would produce **8**.

To distinguish between the two possible pathways (**a** or **b**), we performed a deuterium-labeling study (Scheme 6).¹⁷ Hence, the propargylic ester 7-D4 with 92% deuterium enrichment at the α -carbons (C5 and C6) was subjected to the reaction conditions. As a result, the pentannulated product **8-D4** was obtained with >99% enrichment at C1, whereas the C6 position contained 83% of deuterium. This observation strongly supports **path a** involving a clean 1,2-deuterium shift¹⁸ over **path b**, which

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In summary, a new gold-catalyzed cascade reaction of propargylic esters possessing a ketone moiety to cyclopentenones has been developed. This method allowed for efficient synthesis of various mono-, bis-, and tricyclic cyclopentenones possessing alkyl, aryl, and heterocyclic substituents. ¹³C- and deuterium-labeling studies suggested that this transformation proceeds via a cascade of 1,3-transposition of alkynyl ketone/1,3-acyloxy migration/pentannulation cyclization.

ASSOCIATED CONTENT

Supporting Information

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Experimental procedures, compound characterization data, and crystallographic data (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: vlad@uic.edu.

Notes

The authors declare no competing financial interest.

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REFERENCES

For reviews, see: (a) Fensterbank, L.; Malacria, M. Acc. Chem. Res.
 2014, 47, 953. (b) Kazem Shiroodi, R.; Gevorgyan, V. Chem. Soc. Rev.
 2013, 42, 4991. (c) Shu, X. Z.; Shu, D. X.; Schienebeck, C. M.; Tang, W. P. Chem. Soc. Rev. 2012, 41, 7698. (d) Wang, S. Z.; Zhang, G. Z.; Zhang, L. M. Synlett 2010, 2010, 692. (e) Marco-Contelles, J.; Soriano, E. Chem. - Eur. J. 2007, 13, 1350. Also, see: (g) Rao, W. D.; Koh, M. J.; Li, D.; Hirao, H.; Chan, P. W. H. J. Am. Chem. Soc. 2013, 135, 7926. (f) Garayalde, D.; Gomez-Bengoa, E.; Huang, X. G.; Goeke, A.; Nevado, C. J. Am. Chem. Soc. 2010, 132, 4720.

(2) (a) Prasad, B. A. B.; Yoshimoto, F. K.; Sarpong, R. J. Am. Chem. Soc. **2005**, 127, 12468. (b) Zi, W.; Wu, H.; Toste, F. D. J. Am. Chem. Soc. **2015**, 137, 3225. (c) Hoffmann, M.; Weibel, J. M.; de Fremont, P.; Pale, P.; Blanc, A. Org. Lett. **2014**, 16, 908. Also, see: (d) Pujanauski, B. G.; Prasad, B. A. B.; Sarpong, R. J. Am. Chem. Soc. **2006**, 128, 6786.

(3) For recent examples on gold catalysis, see: (a) Dorel, R.; Echavarren, A. M. Chem. Rev. 2015, ASAP (DOI: 10.1021/cr500691k).
(b) Furstner, A. Chem. Soc. Rev. 2009, 38, 3208. (c) Liu, L. P.; Hammond, G. B. Chem. Soc. Rev. 2012, 41, 3129. (d) Wang, Y. M.; Lackner, A. D.; Toste, F. D. Acc. Chem. Res. 2014, 47, 889. (e) Friend, C. M.; Hashmi, A. S. K. Acc. Chem. Res. 2014, 47, 729. (f) Xie, J.; Pan, C.; Abdukader, A.; Zhu, C. Chem. Soc. Rev. 2014, 43, 5245. (g) Cao, Z. P.; Gagosz, F. Angew. Chem., Int. Ed. 2013, 52, 9014. (h) Kazem Shiroodi, R.; Koleda, O.; Gevorgyan, V. J. Am. Chem. Soc. 2014, 136, 13146.
(i) Hofer, M.; Gomez-Bengoa, E.; Nevado, C. Organometallics 2014, 33, 1328.

(4) For a gold-catalyzed pentannulation reaction of alkynylspirocyclopropanes, see: Barluenga, J.; Sigueiro, R.; Vicente, R.; Ballesteros, A.; Tomas, M.; Rodriguez, M. A. *Angew. Chem., Int. Ed.* **2012**, *51*, 10377.

(5) (a) Sromek, A. W.; Kel'in, A. V.; Gevorgyan, V. Angew. Chem., Int. Ed. 2004, 43, 2280. (b) Schwier, T.; Sromek, A. W.; Yap, D. M. L.; Chernyak, D.; Gevorgyan, V. J. Am. Chem. Soc. 2007, 129, 9868. For a

gold-catalyzed version, see: (c) Hoffmann, M.; Miaskiewicz, S.; Weibel, J. M.; Pale, P.; Blanc, A. Beilstein J. Org. Chem. **2013**, *9*, 1774.

(6) For a review, see ref 1b. Also, see: (a) Kazem Shiroodi, R.; Dudnik, A. S.; Gevorgyan, V. *J. Am. Chem. Soc.* **2012**, *134*, 6928. (b) Dudnik, A. S.; Schwier, T.; Gevorgyan, V. Org. Lett. **2008**, *10*, 1465. (c) Gronnier, C.; Boissonnat, G.; Gagosz, F. Org. Lett. **2013**, *15*, 4234. (d) de Haro, T.; Gomez-Bengoa, E.; Cribiu, R.; Huang, X. G.; Nevado, C. Chem. - Eur. J. **2012**, *18*, 6811.

(7) Dudnik, A. S.; Gevorgyan, V. Angew. Chem., Int. Ed. 2007, 46, 5195. Dudnik, A. S.; Sromek, A. W.; Rubina, M.; Kim, J. T.; Kel'in, A. V.; Gevorgyan, V. J. Am. Chem. Soc. 2008, 130, 1440.

(8) See Supporting Information for details.

(9) Istrate, F. M.; Gagosz, F. Org. Lett. 2007, 9, 3181.

(10) The structure was confirmed by X-ray crystallography. See Supporting Information for details.

(11) The hydroxylated cyclopentenone cores are present in a number of natural products. For antibiotic tylopilusins, see: (a) Fukuda, T.; Nagai, K.; Tomoda, H. J. Nat. Prod. 2012, 75, 2228. For the anti-HIV and antianagiogenesis terpestacin family, see: (b) Oka, M.; Iimura, S.; Tenmyo, O.; Sawada, Y.; Sugawara, M.; Ohkusa, N.; Yamamoto, H.; Kawano, K.; Hu, S. L.; Fukagawa, Y.; Oki, T. J. Antibiot. 1993, 46, 367. (c) Jung, H. J.; Lee, H. B.; Kim, C. J.; Rho, J.-R.; Shin, J.; Kwon, H. J. J. Antibiot. 2003, 56, 492. For importance of this core in cosmetics, see: (d) Belsito, D.; Bickers, D.; Bruze, M.; Calow, P.; Dagli, M. L.; Dekant, W.; Fryer, A. D.; Greim, H.; Miyachi, Y.; Saurat, J. H.; Sipes, I. G.; Panel, R. E. Food Chem. Toxicol. 2012, 50, S517.

(12) The cyclobutane-containing substrate did not react under these reaction conditions. The substrates possessing the cyclohexane group produced an inseparable mixture of products.

(13) For a review, see: Obradors, C.; Echavarren, A. M. *Acc. Chem. Res.* **2014**, 47, 902.

(14) For 1,3-transposition of ynones, see: Kazem Shiroodi, R.; Soltani, M.; Gevorgyan, V. J. Am. Chem. Soc. **2014**, *136*, 9882.

(15) The proposed intermediate B was not observed in the course of the reaction, which supports its formation at a rate-limiting event.

(16) For recent reviews on Nazarov cyclization, see: (a) Vaidya, T.; Eisenberg, R.; Frontier, A. J. ChemCatChem 2011, 3, 1531. (b) Tius, M. A. Chem. Soc. Rev. 2014, 43, 2979. For examples on the Au-catalyzed Nazarov cyclization, see: (c) Shi, X. D.; Gorin, D. J.; Toste, F. D. J. Am. Chem. Soc. 2005, 127, 5802. (d) Jin, T.; Yamamoto, Y. Org. Lett. 2008, 10, 3137. (e) Lemiere, G.; Gandon, V.; Cariou, K.; Hours, A.; Fukuyama, T.; Dhimane, A. L.; Fensterbank, L.; Malacria, M. J. Am. Chem. Soc. 2009, 131, 2993. (f) Lin, C. C.; Teng, T. M.; Tsai, C. C.; Liao, H. Y.; Liu, R. S. J. Am. Chem. Soc. 2008, 130, 16417. (g) Zhang, L. M.; Wang, S. Z. J. Am. Chem. Soc. 2006, 128, 1442.

(17) (a) Jones, W. D. Acc. Chem. Res. **2003**, 36, 140. (b) Singleton, D. A.; Hang, C. J. Org. Chem. **2000**, 65, 895. (c) Horino, Y.; Yamamoto, T.; Ueda, K.; Kuroda, S.; Toste, F. D. J. Am. Chem. Soc. **2009**, 131, 2809.

(18) The origin of this apparent inverse kinetic isotope effect is not clear at this point.

(19) For substantial D-label scrambling observed in Au-catalyzed processes involving prototropic isomerization, see, for example: (a) Xia, Y. Z.; Dudnik, A. S.; Li, Y. H.; Gevorgyan, V. Org. Lett. 2010, 12, 5538.
(b) Gorin, D. J.; Dube, P.; Toste, F. D. J. Am. Chem. Soc. 2006, 128, 14480. (c) Davies, P. W.; Martin, N.; Spencer, N. Beilstein J. Org. Chem. 2011, 7, 839.